

## 三个不同穿插的锌配合物的合成、结构和热动力学分析

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## Synthesis, Structure and Thermodynamics/Kinetics Analysis of Three Different Interpenetrating Zinc(II) Coordination Architectures

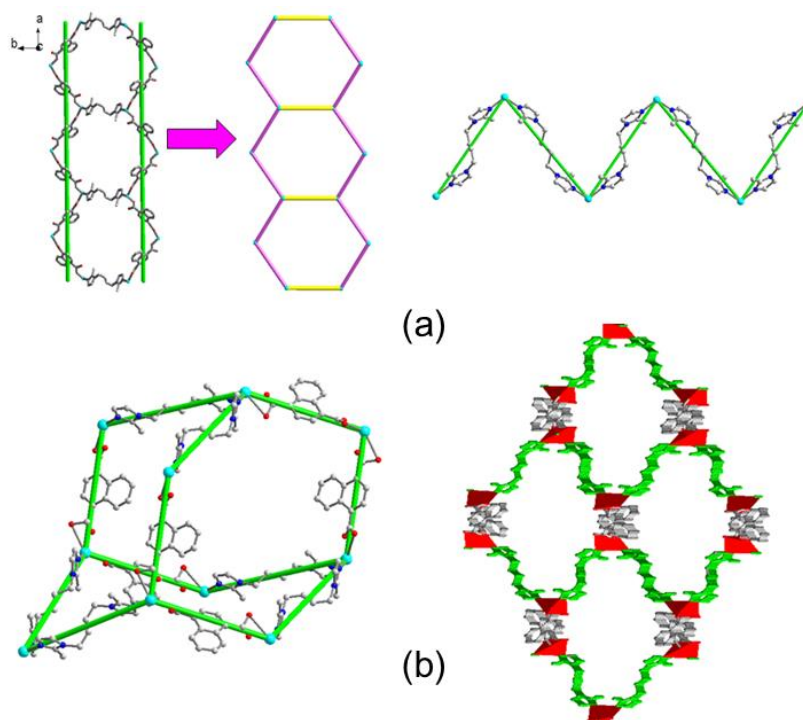
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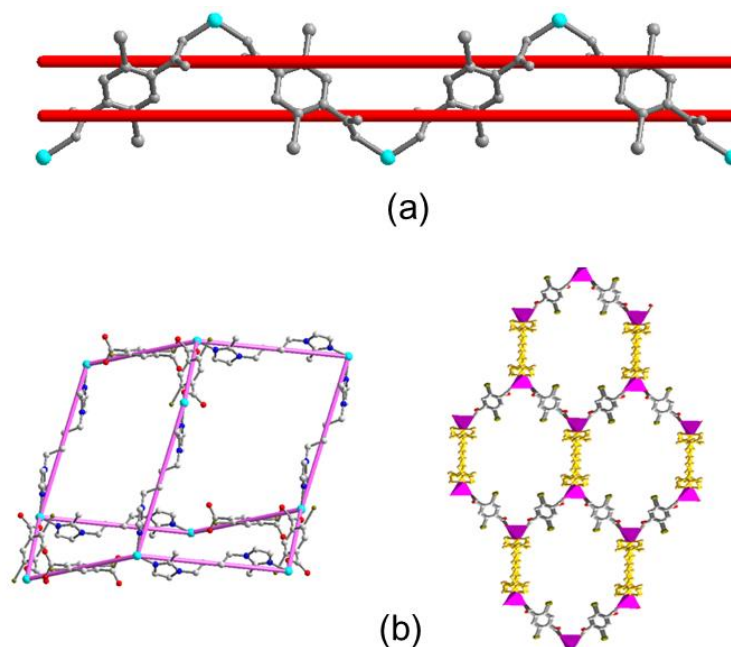
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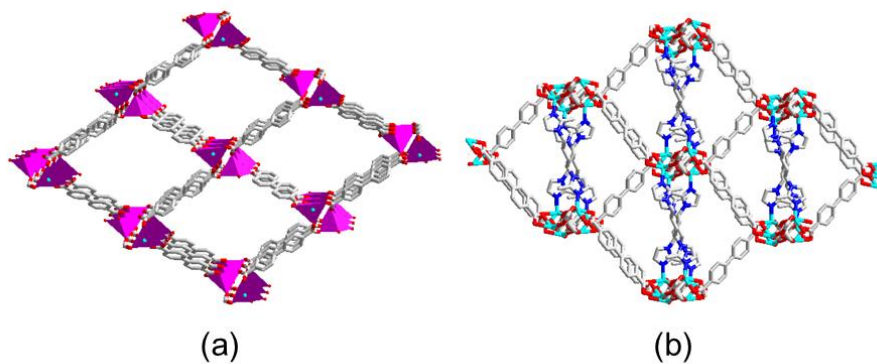
ZHOU Chun-Sheng, Email: slzhoucs@126.com.cn.



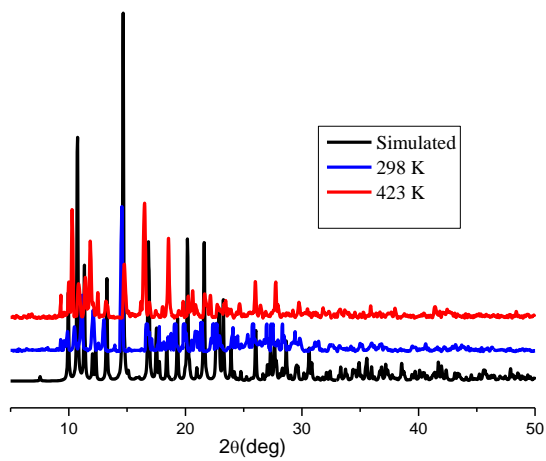
**Fig.S1 (a) two kinds of  $[\text{Zn}(1,4\text{-ndc})]_n$  zigzag chains and  $[\text{Zn}(\text{bib})]_n$  zigzag chain; (b) left: A single adamantanoid cage of 1, right: A single dia network**



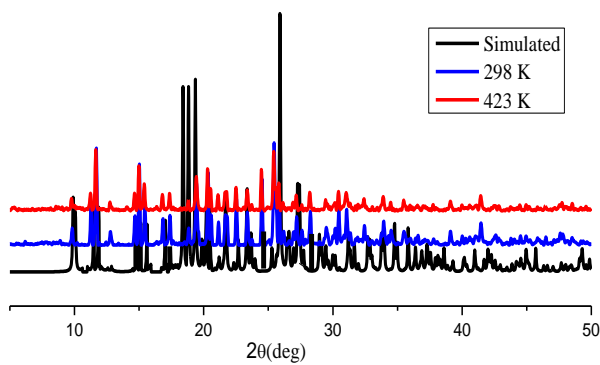
**Fig.S2 (a) View of the *meso*-helical chain in 2; (b) left: A single adamantanoid cage of 2, right: A single 3D dia network**



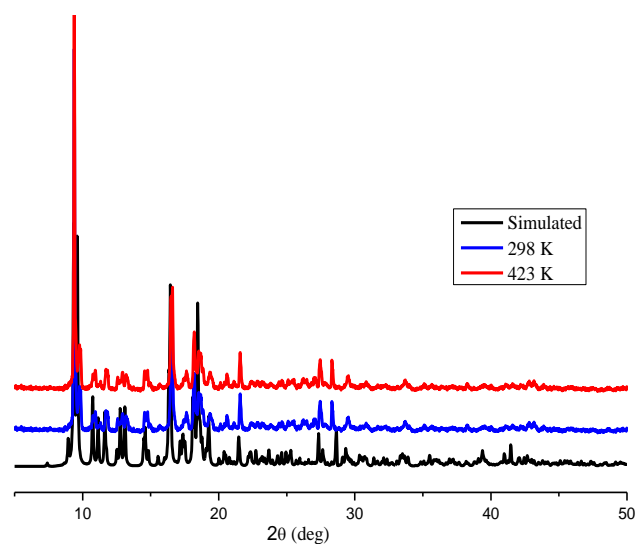
**Fig.S3 (a) 3D framework formed by only Zn(II) and 4,4'-bpdca<sup>2-</sup> in 3; (b) 3D framework of 3**



(a)



(b)



(c)

Fig.S4 PXR D patterns for 1(a), 2(b) and 3(c) recorded N<sub>2</sub> atmosphere

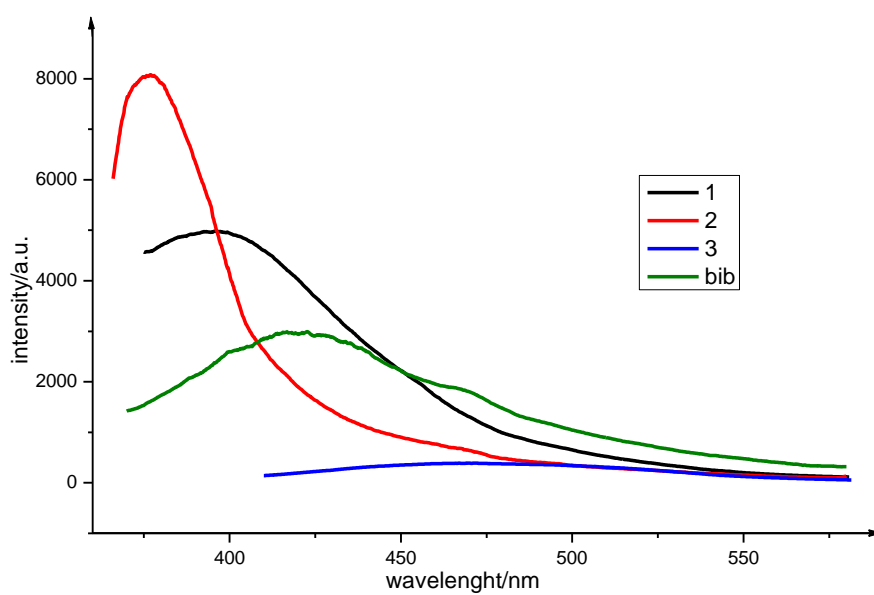


Fig.S5 Luminescences of free ligand bib and complexes 1–3

Table S1 Crystal data and structure refinements for 1–3

Complex	1	2	3
Empirical formula	C <sub>48</sub> H <sub>46</sub> Zn <sub>2</sub> N <sub>8</sub> O <sub>8</sub>	C <sub>20</sub> H <sub>20</sub> ZnBr <sub>2</sub> N <sub>4</sub> O <sub>7</sub>	C <sub>40</sub> H <sub>34</sub> Zn <sub>2</sub> N <sub>4</sub> O <sub>9</sub>
Formula weight	993.71	653.59	845.45
Temperature/K	296(2)	296(2)	296(2)
Crystal system	Monoclinic	orthorhombic	Monoclinic
Space group	<i>C2/c</i>	<i>Pnna</i>	<i>C2/c</i>

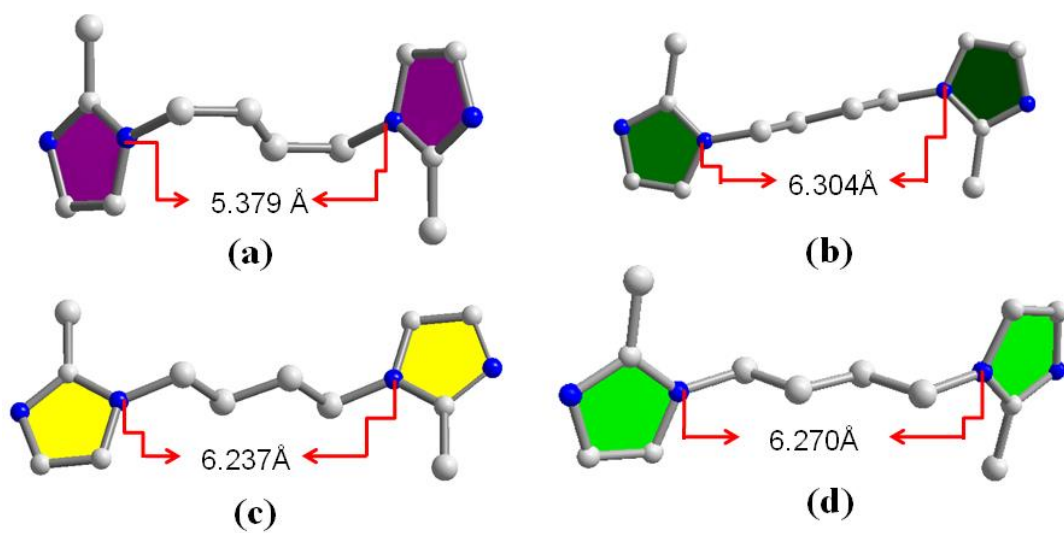
$a/\text{\AA}$	20.273(5)	17.862(4)	24.242(4)
$b/\text{\AA}$	17.762(5)	13.888(3)	15.713(2)
$c/\text{\AA}$	16.121(5)	10.086(2)	10.7475(16)
$\alpha/^\circ$	90	90	90
$\beta/^\circ$	125.668(5)	90	101.581(3)
$\gamma/^\circ$	90	90	90
$V/\text{\AA}^3$	4716(6)	2502.1(9)	4010.6(10)
$Z$	4	4	4
Calculated density ( $\text{mg m}^{-3}$ )	1.467	1.735	1.400
$R_{\text{int}}$	0.1136	0.1041	0.0664
GOF on $F^2$	1.267	0.676	1.012
$R_1^a[I > 2\sigma(I)]$	0.1396	0.0832	0.0575
$wR_2^b[I > 2\sigma(I)]$	0.3580	0.1869	0.1599

$$^a R_1 = \sum(|F_o| - |F_c|) / \sum |F_o|, \quad ^b wR_2 = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2]^{1/2}.$$

**Table S2 Selected bond distances (Å) and angles (°) for 1–3.**

Complex 1			
Zn(1)-O(5)	1.916(8)	Zn(1)-O(2)	2.002(7)
Zn(1)-N(4)	2.024(9)	Zn(1)-N(2)	2.039(8)
O(5)-Zn(1)-O(2)	98.6(3)	O(5)-Zn(1)-N(4)	117.4(4)
O(2)-Zn(1)-N(4)	120.4(3)	O(5)-Zn(1)-N(2)	113.4(4)
O(2)-Zn(1)-N(2)	99.4(3)	N(4)-Zn(1)-N(2)	106.5(4)
Complex 2			
Zn(1)-O(2)#1	1.928(7)	Zn(1)-O(2)	1.928(6)
Zn(1)-N(1)	2.011(7)	Zn(1)-N(1)#1	2.011(7)
O(2)#1-Zn(1)-O(2)	114.3(5)	O(2)#1-Zn(1)-N(1)	104.4(3)
O(2)#1-Zn(1)-N(1)#1	112.1(3)	O(2)-Zn(1)-N(1)	112.1(3)
O(2)-Zn(1)-N(1)#1	104.4(3)	N(1)-Zn(1)-N(1)#1	109.7(5)
Complex 3			
Zn(1)-O(2)	1.945(4)	Zn(1)-O(4)	1.958(4)
Zn(1)-N(2)	1.991(4)	Zn(1)-O(3)	2.002(3)
O(2)-Zn(1)-O(4)	105.62(19)	O(2)-Zn(1)-N(2)	137.22(19)
O(4)-Zn(1)-N(2)	98.41(19)	O(2)-Zn(1)-O(3)	103.30(18)
O(4)-Zn(1)-O(3)	113.74(16)	N(2)-Zn(1)-O(3)	98.46(16)

Symmetry transformations used to generate equivalent atoms: **For 1**, #1  $x+1/2, -y+1/2, z+1/2$  #2  $-x+1/2, -y-1/2, -z+2$  #3  $-x+1/2, -y-1/2, -z+1$  #4  $x-1/2, -y+1/2, z-1/2$ ; **For 2**, #1  $x, -y+1/2, -z+1/2$  #2  $-x+1, -y, -z$  #3  $-x, -y+1, -z$ ; **For 3**, #1  $-x+1/2, -y+1/2, -z+1$  #2  $-x+1, y, -z+1/2$  #3  $-x, -y+1, -z$  #4  $-x+1/2, -y-1/2, -z+1$



**Scheme S1** A variety of spatial configuration of bib in 1–3