

阴离子功能化大孔树脂用于二氧化硫的多位点高效捕集

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Ultra-high SO₂ Capture by Anion Functionalized Resins through Multiple-Site Adsorption

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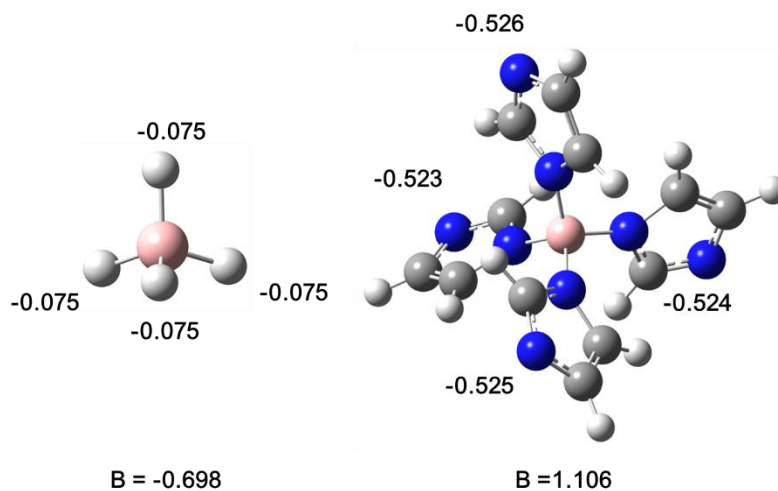
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Table S1 Element analysis of the anion functionalized resins [IRA-900][An].

Material	C (%)	H (%)	N (%)	Anion sites (mmol g ⁻¹) ^a
[IRA-900][Cl]	63.46	8.40	5.35	3.82
[IRA-900][B(Im) ₄]	58.86	7.57	23.43	1.86
[IRA-900][Triz]	63.66	8.84	19.95	3.56
[IRA-900][Tetz]	59.07	8.45	25.89	3.70

^a Calculated by the nitrogen element content.**Table S2 The comparison of SO₂ adsorption capacities by [IRA-900][B(Im)₄] with other selected sorbents at 0.1 bar.**

Entry	Material	Temperature (°C)	Pressure (kPa)	SO ₂ adsorption capacity (mmol g ⁻¹)	Reference
1	[IRA-900][B(Im) ₄]	20	10.13	10.62	This work
2	SIFSIX-1-Cu	25	10.13	8.74	S1
3	SIFSIX-2-Cu-i	25	10.13	6.01	S1
4	SIFSIX-3-Ni	25	10.13	2.55	S1
5	NOTT-202a	10	10.13	6.2	S2
6	Pillar[5]arene 4Tetz	20	10.13	8.2	S3
7	[TMEDAMR]Cl	30	101.3	6.40	S4
8	[P ₆₆₆₁₄][Im]	20	10.13	3.76	S5
9	[Emim][SCN]	20	10.13	5.83	S6

**Fig. S1 NBO charges of atoms on [BH₄]⁻ (left) as well as electronegative nitrogens and boron on [B(Im)₄]⁻ (right).**

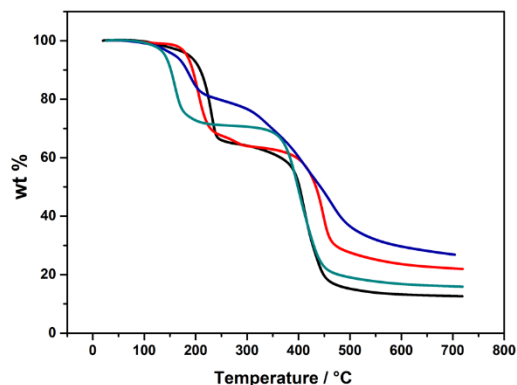


Fig. S2 Thermostabilities of the anion functionalized macro-porous resins by TGA. [IRA-900][Cl] (black), [IRA-900][Triz] (green), [IRA-900][Tetz] (red), [IRA-900][B(Im)₄] (blue).

All the heating rates were controlled as 10 °C min⁻¹.

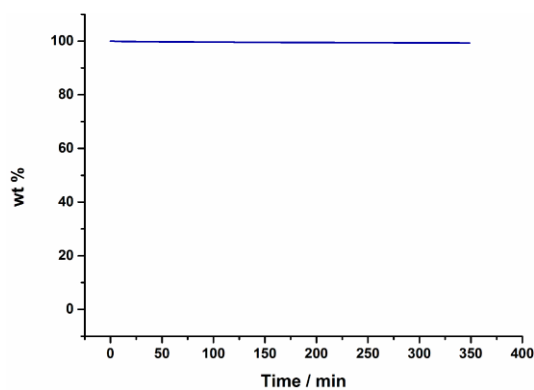


Fig. S3 Thermostability of the anion functionalized macro-porous resin [IRA-900][B(Im)₄] by TGA.

Temperature was controlled at 70 °C at N₂ atmosphere for 6 h.

References

- S1 Cui, X. L.; Yang, Q. W.; Yang, L. F.; Krishna, R.; Zhang, Z. G.; Bao, Z. B.; Wu, H.; Ren, Q. L.; Zhou, W.; Chen, B. L.; Xing, H. B. *Adv. Mater.* **2017**, *29* (28). doi: 10.1002/adma.201606929
- S2 Yang, S. H.; Liu, L. F.; Sun, J. L.; Thomas, K. M.; Davies, A. J.; George, M. W.; Blake, A. J.; Hill, A. H.; Fitch, A. N.; Tang, C. C.; Schroder, M. *J. Am. Chem. Soc.* **2013**, *135* (13), 4954. doi: 10.1021/ja401061m
- S3 Lin, W. J.; Zhou, X. Y.; Cai, J. S.; Chen, K. H.; He, X.; Kong, X. Q.; Li, H. R.; Wang, C.M. *Chem-Eur. J.* **2017**, *23*, 14143. doi: 10.1002/chem.201703007
- S4 Lee, H. J.; Lee, K. I.; Kim, M.; Suh, Y. W.; Kim, H. S.; Lee, H. *Acs Sustain. Chem. Eng.* **2016**, *4* (4), 2012. doi: 10.1021/acssuschemeng.5b01325
- S5 Wang, C. M.; Cui, G. K.; Luo, X. Y.; Xu, Y. J.; Li, H. R.; Dai, S. *J. Am. Chem. Soc.* **2011**, *133* (31), 11916. doi: 10.1021/ja204808h
- S6 Wang, C. M.; Zheng, J. J.; Cui, G. K.; Luo, X. Y.; Guo, Y.; Li, H. R. *Chem. Commun.* **2013**, *49* (12), 1166. doi: 10.1039/c2cc37092a