

## 负载型Rh基催化剂上的CO加氢制乙醇反应：载体效应

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## CO Hydrogenation to Ethanol over Supported Rh-Based Catalyst: Effect of the Support

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**Table S1 Performance of CO hydrogenation over various catalysts <sup>a</sup>.**

Catalyst	$X_{CO}^{b/\%}$	Selectivity (%)			Alcohol distribution (w, %)				
		ROH <sup>c</sup>	CH <sub>n</sub> <sup>d</sup>	CO <sub>2</sub>	MeOH	EtOH	PrOH	BuOH	C <sub>5+</sub> -OH
RhFe/CeO <sub>2</sub>	5.1	25.6	71.3	3.1	60.2	30.1	7.8	1.5	0.4
RhFe/ZrO <sub>2</sub>	7.2	27.8	70.7	1.6	52.5	39.4	6.5	1.3	0.3
RhFe/TiO <sub>2</sub>	18.2	33.2	62.4	4.4	15.6	74.7	4.3	5.2	0.2
Fe/TiO <sub>2</sub>	1.4	–	> 99.9	–	–	–	–	–	–

<sup>a</sup> Reaction conditions: H<sub>2</sub>/CO = 2, 250 °C, 2 MPa, 40 mL·min<sup>-1</sup>, TOS = 48 h.

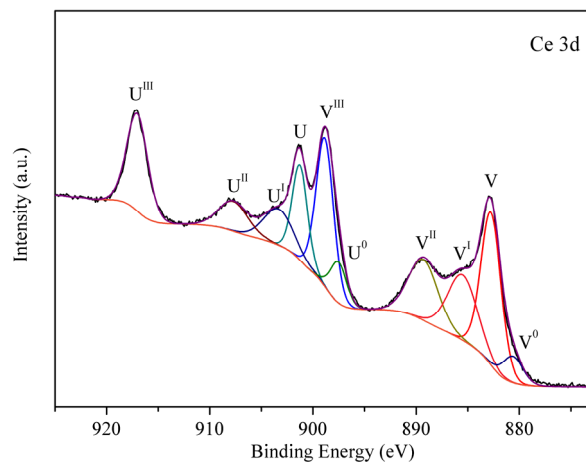
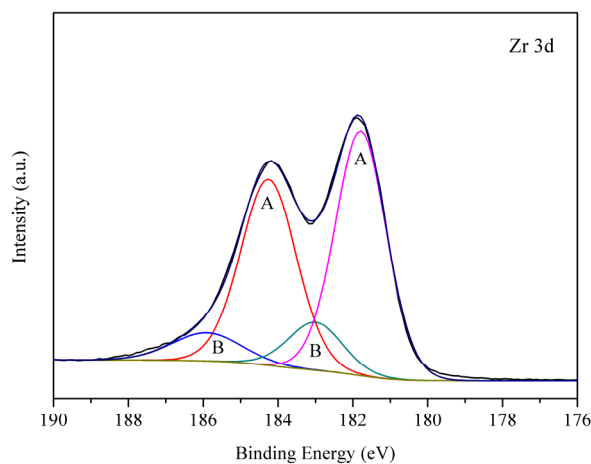
**Table S2 ICP-OES and XPS results, surface species content for different catalysts.**

Catalyst	Loading (w, %)		Surface atomic percentage <sup>a</sup>		Surface species content <sup>b</sup> (w, %)	
	Rh	Fe	Rh <sup>0</sup>	Rh <sup>+</sup>	Rh <sup>0</sup>	Rh <sup>+</sup>
RhFe/CeO <sub>2</sub>	2.0	3.9	307.1(77.9%)	308.8(22.1%)	0.31	0.09
RhFe/ZrO <sub>2</sub>	2.3	3.9	307.2(66.0%)	308.9(34.0%)	0.37	0.19
RhFe/TiO <sub>2</sub>	2.1	4.2	307.1(52.4%)	308.9(47.6%)	0.54	0.49

<sup>a</sup> Estimated by XPS spectrum fitting result of Rh 3d in different catalyst. <sup>b</sup> Estimated by actual loading, metal dispersion and the percentage of Rh<sup>0</sup> and Rh<sup>+</sup>.

**Table S3 Surface atomic ratio estimated by XPS for different catalysts.**

Catalyst	Ratio
	O <sub>II</sub> /(O <sub>I</sub> + O <sub>II</sub> + O <sub>III</sub> )
RhFe/CeO <sub>2</sub>	0.21
RhFe/ZrO <sub>2</sub>	0.22
RhFe/TiO <sub>2</sub>	0.40

**Fig. S1 Ce 3d region of RhFe/CeO<sub>2</sub>.****Fig. S2 Zr 3d region of RhFe/ZrO<sub>2</sub>.**

The peaks of Ce  $3d_{5/2}$  and Ce  $3d_{3/2}$  in Fig. S1 labeled as V, V<sup>II</sup>, V<sup>III</sup> and U, U<sup>II</sup>, U<sup>III</sup> corresponded to unreduced Ce<sup>4+</sup>, V<sup>0</sup>, V<sup>I</sup>, U<sup>0</sup> and U<sup>I</sup> were features of Ce<sup>3+</sup>. Partial reduction of Ce<sup>4+</sup> and formation of oxygen vacancies on RhFe/CeO<sub>2</sub> could be achieved during XPS pretreatment in a vacuum, in accordance with the viewpoint of Fierro *et al.*<sup>2</sup>. On Fig. S2, the spectrum of Zr 3d was deconvoluted into two parts (A and B), which were characteristic for unreduced Zr<sup>4+</sup><sup>3</sup>. Thus, it was inferred that a little bit of O<sub>II</sub> in RhFe/ZrO<sub>2</sub> should be due to the interaction between metal and support.

## References

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