

含氮金属有机框架衍生的铜基催化剂电催化还原二氧化碳

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Cu-based Catalyst Derived from Nitrogen-containing Metal Organic Frameworks for Electroreduction of CO₂

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NMR 产物标定

核磁共振氢谱(^1H NMR)在 600 MHz 下由 Agilent Technologies 的 DD2-600 液体超导核磁共振谱仪得到。取含有一定浓度的内标物(DMSO) 50 μL , 100 μL D_2O 加入到 400 μL 电解液溶液中制得核磁样。在 ^1H NMR 波谱测试中采用水峰压制。对峰面积进行积分, 通过产物的 NMR 标线: 产物与内标物的峰面积之比 $S_1 : S_2$ 与产物的浓度曲线可以计算出溶质产物的浓度。产物的 NMR 标线及其线性方程如下图 S1 和表 S1 所示。样品在电解质中的 ^1H NMR 波谱如图 S1 所示。

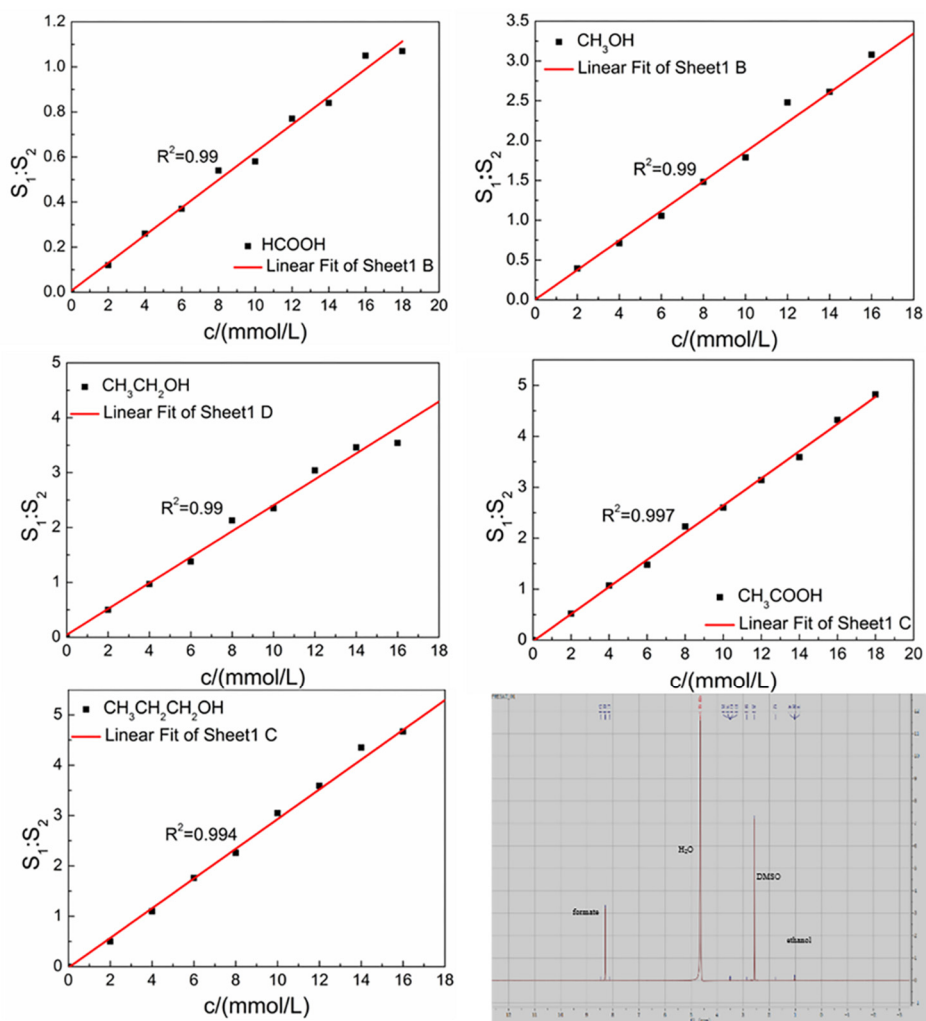


图 S1 各种产物的核磁标线及样品在电解质中的核磁共振氢谱

Fig. S1 The NMR standard lines of each products and ^1H NMR spectrum of samples in electrolyte.

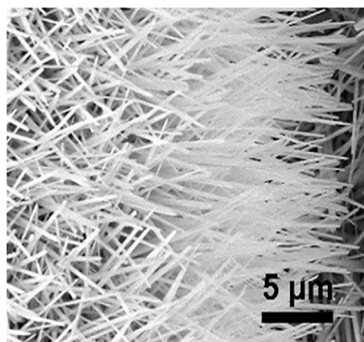


图 S2 $\text{Cu}(\text{OH})_2$ NA/CF 的 SEM 图

Fig. S2 SEM image of $\text{Cu}(\text{OH})_2$ NA/CF.

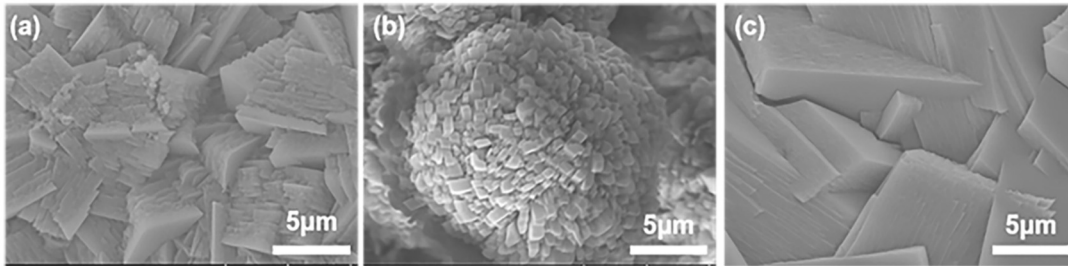


图 S3 三种溶剂反应 72 h 所得的 Cu-NBDC 的 SEM 图

Fig. S3 SEM images of Cu-NBDC reacted for 72 h with three kinds of solvent.

(a) 20 mL DMF; (b) 10 mL DMF, 10 mL H₂O and (c) 10 mL DMF, 10 mL EtOH.

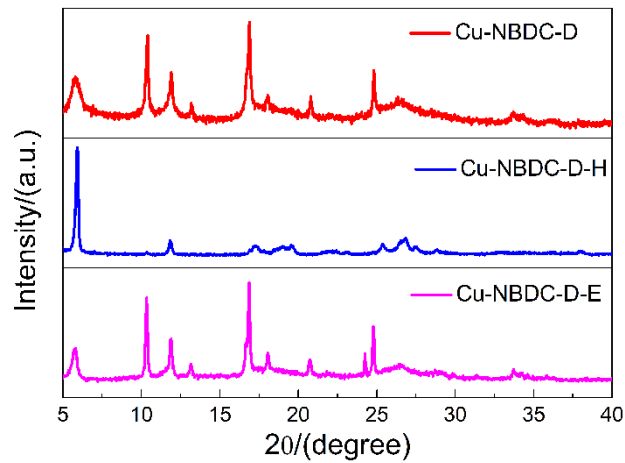


图 S4 三种 Cu-NBDC 粉末的 XRD 谱图

Fig. S4 XRD patterns of three kinds of Cu-NBDC powders.

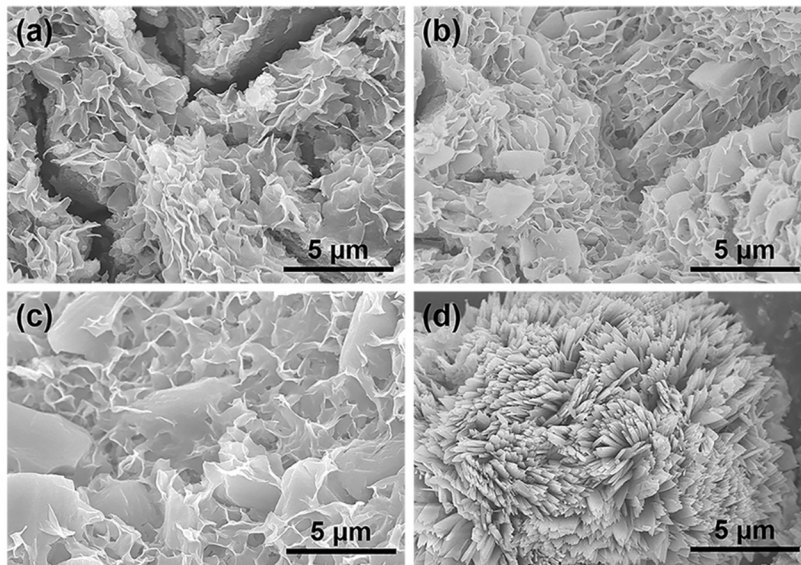


图 S5 不同溶剂比制备的 Cu-NBDC 的 SEM 图

Fig. S5 SEM images of Cu-NBDC prepared by varying the solvent ratio.

(a) $V(\text{DMF}) : V(\text{H}_2\text{O}) = 100 : 1$; (b) $V(\text{DMF}) : V(\text{H}_2\text{O}) = 20 : 1$; (c) $V(\text{DMF}) : V(\text{H}_2\text{O}) = 10 : 1$; (d) $V(\text{DMF}) : V(\text{H}_2\text{O}) = 5 : 1$.

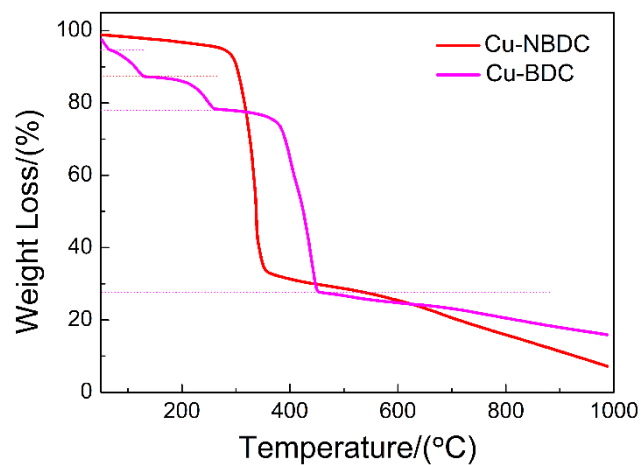


图 S6 Cu-NBDC 与 Cu-BDC 的热重分析图

Fig. S6 TGA for Cu-NBDC and Cu-BDC.

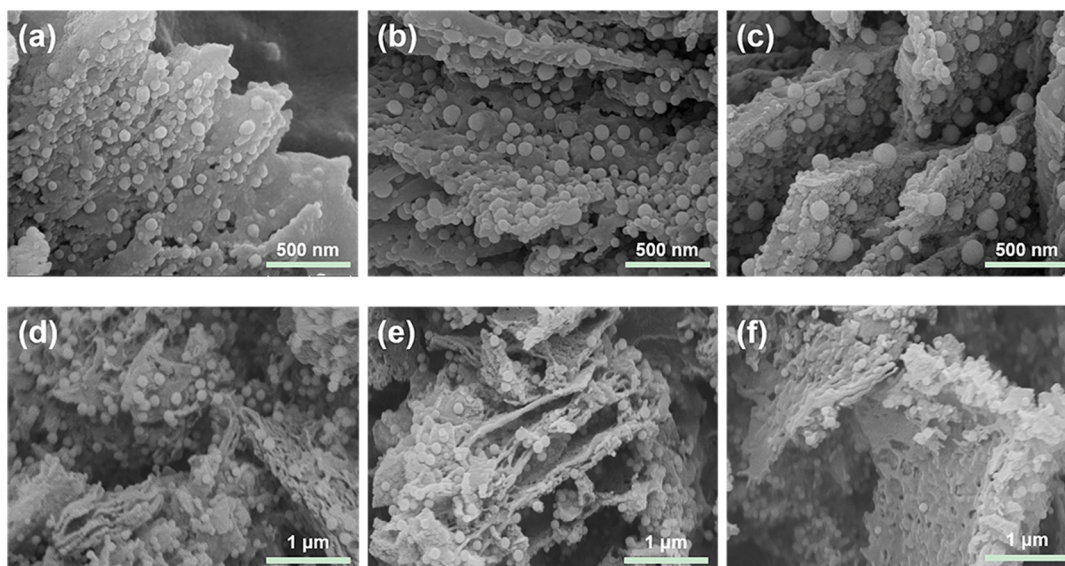


图 S7 不同温度下退火的 $\text{Cu}_2\text{O}/\text{Cu}@\text{NC}$ 与 $\text{Cu}_2\text{O}/\text{Cu}@\text{C}$ 的 SEM 图

Fig. S7 SEM images of $\text{Cu}_2\text{O}/\text{Cu}@\text{NC}$ and $\text{Cu}_2\text{O}/\text{Cu}@\text{C}$ annealed at different temperatures.

(a) $\text{Cu}_2\text{O}/\text{Cu}@\text{NC}-400$; (b) $\text{Cu}_2\text{O}/\text{Cu}@\text{NC}-600$; (c) $\text{Cu}_2\text{O}/\text{Cu}@\text{NC}-800$; (d) $\text{Cu}_2\text{O}/\text{Cu}@\text{C}-400$; (e) $\text{Cu}_2\text{O}/\text{Cu}@\text{C}-600$ and (f) $\text{Cu}_2\text{O}/\text{Cu}@\text{C}-800$.

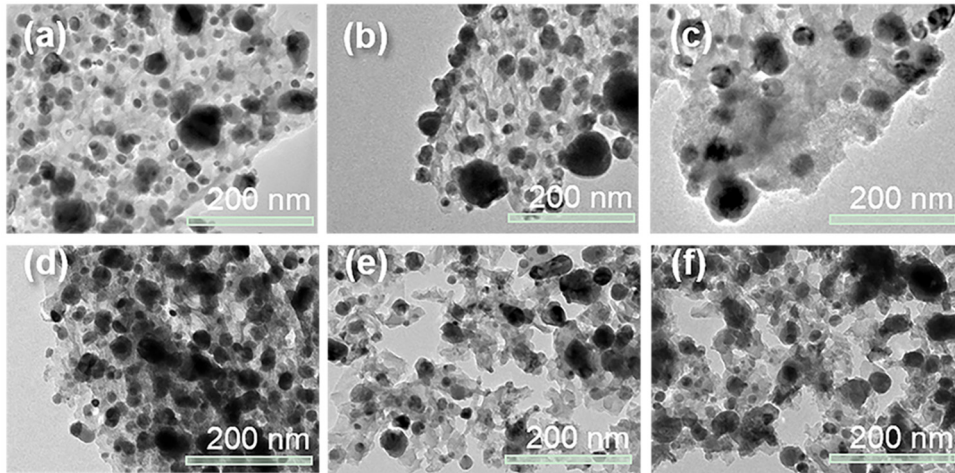


图 S8 不同温度下退火的 $\text{Cu}_2\text{O}/\text{Cu}@\text{NC}$ 与 $\text{Cu}_2\text{O}/\text{Cu}@\text{C}$ 的 TEM 图

Fig. S8 TEM images of $\text{Cu}_2\text{O}/\text{Cu}@\text{NC}$ and $\text{Cu}_2\text{O}/\text{Cu}@\text{C}$ annealed at different temperatures.

(a) $\text{Cu}_2\text{O}/\text{Cu}@\text{NC}-400$; (b) $\text{Cu}_2\text{O}/\text{Cu}@\text{NC}-600$; (c) $\text{Cu}_2\text{O}/\text{Cu}@\text{NC}-800$; (d) $\text{Cu}_2\text{O}/\text{Cu}@\text{C}-400$; (e) $\text{Cu}_2\text{O}/\text{Cu}@\text{C}-600$ and (f) $\text{Cu}_2\text{O}/\text{Cu}@\text{C}-800$.

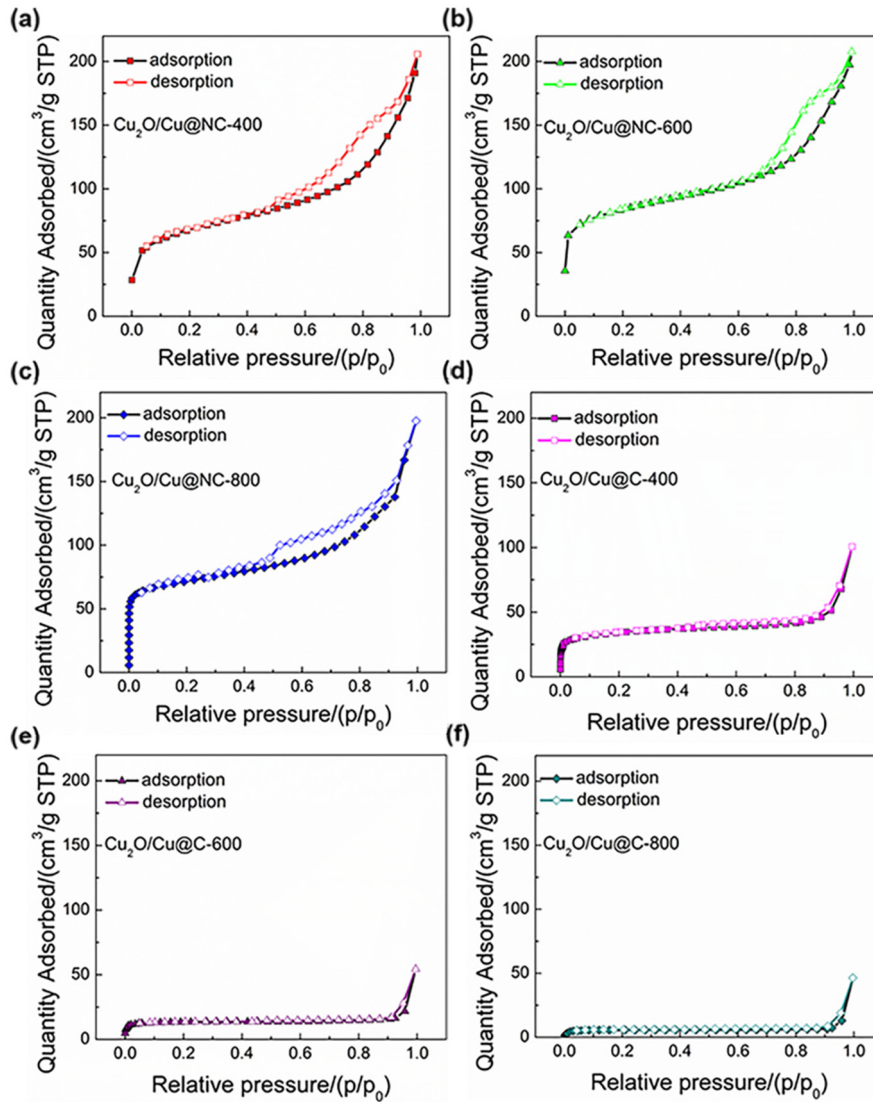


图 S9 不同温度退火的 $\text{Cu}_2\text{O}/\text{Cu}@\text{NC}$ 与 $\text{Cu}_2\text{O}/\text{Cu}@\text{C}$ 的氮气吸-脱附曲线

Fig. S9 N_2 adsorption-desorption isotherms of $\text{Cu}_2\text{O}/\text{Cu}@\text{NC}$ and $\text{Cu}_2\text{O}/\text{Cu}@\text{C}$ annealed at different temperature.

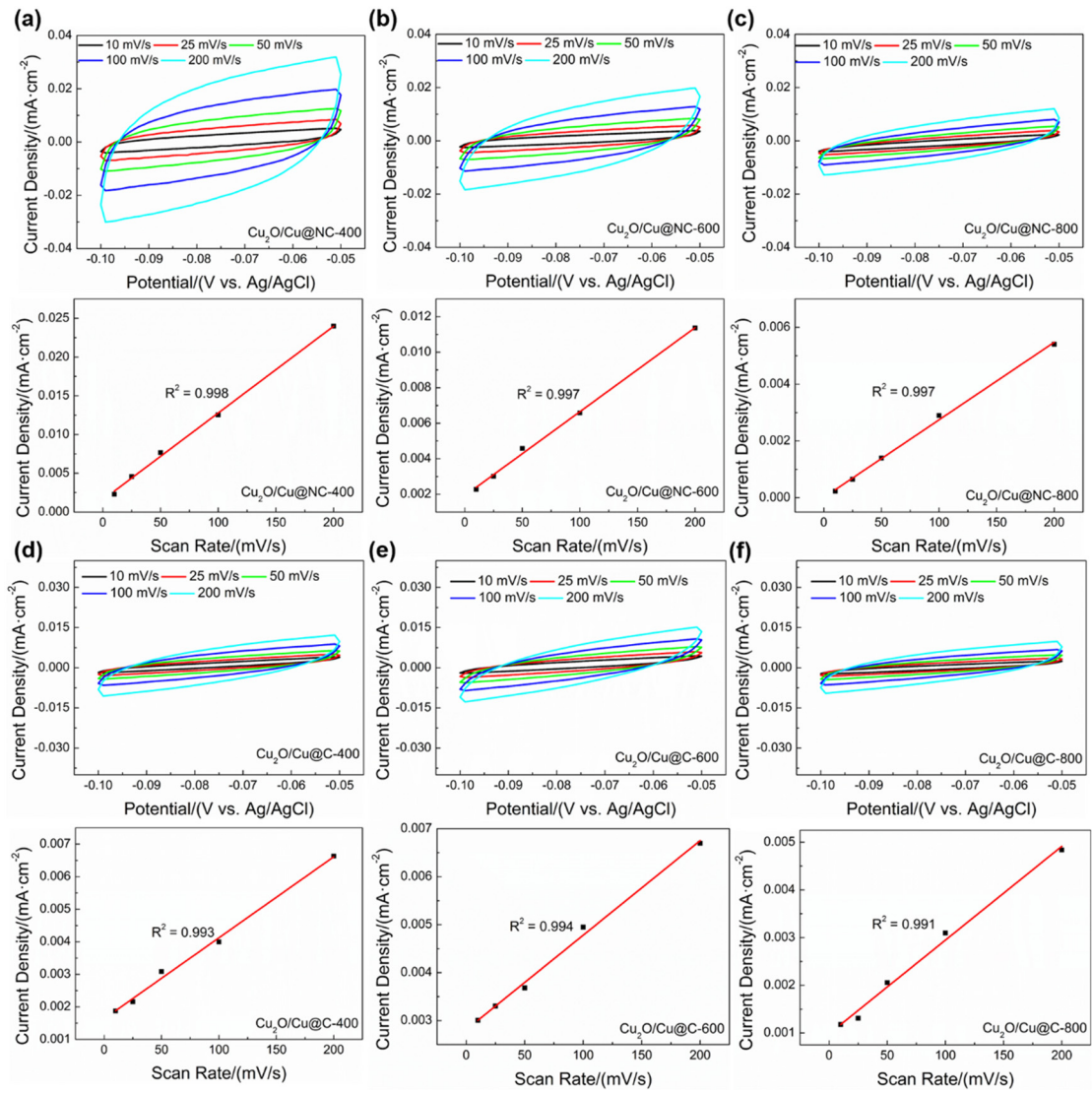


图 S10 测定不同样品的电化学比表面积

Fig. S10 Determining Electrochemical specific surface area of different samples.

(a) $\text{Cu}_2\text{O}/\text{Cu}@NC-400$; (b) $\text{Cu}_2\text{O}/\text{Cu}@NC-600$; (c) $\text{Cu}_2\text{O}/\text{Cu}@NC-800$; (d) $\text{Cu}_2\text{O}/\text{Cu}@C-400$; (e) $\text{Cu}_2\text{O}/\text{Cu}@C-600$ and (f) $\text{Cu}_2\text{O}/\text{Cu}@C-800$.

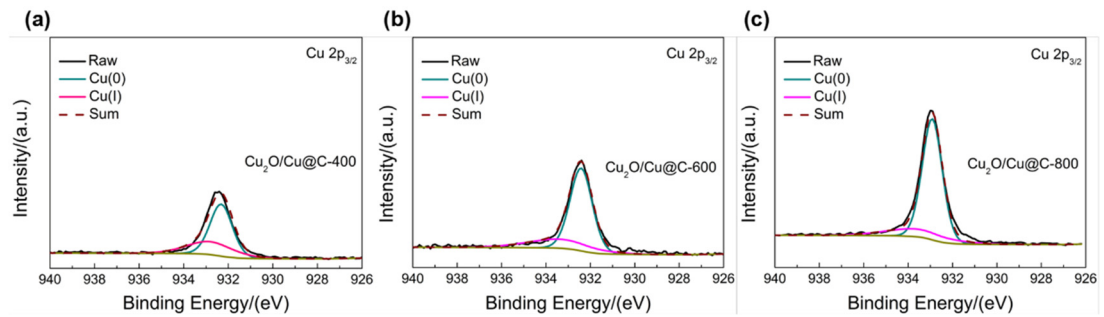


图 S11 $\text{Cu } 2p_{3/2}$ 的高分辨 XPS 光谱图

Fig. S11 High-resolution XPS spectra of $\text{Cu } 2p_{3/2}$.

(a) $\text{Cu}_2\text{O}/\text{Cu}@C-400$; (b) $\text{Cu}_2\text{O}/\text{Cu}@C-600$ and $\text{Cu}_2\text{O}/\text{Cu}@C-800$.

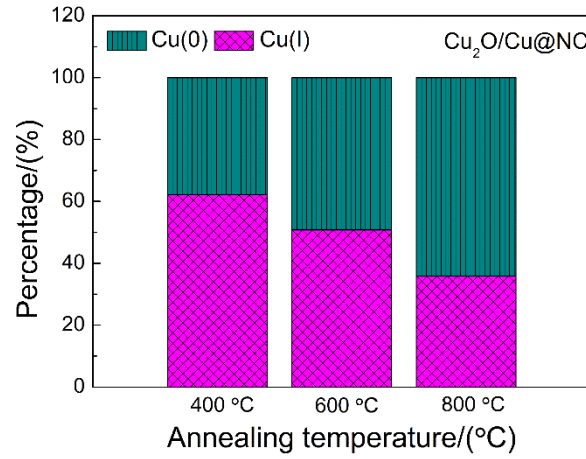


图 S12 不同退火温度下样品中的 Cu/Cu₂O 含量百分比

Fig. S12 Cu/Cu₂O content for Cu₂O/Cu@NC annealed at different temperatures.

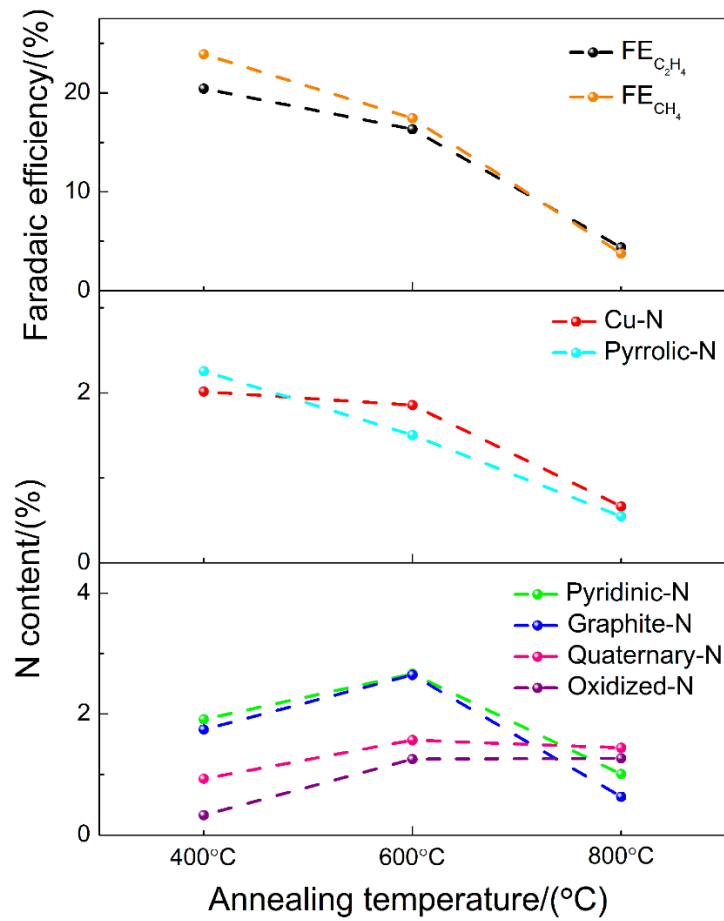


图 S13 Cu₂O/Cu@NC 中 N 含量函数性与乙烯和甲烷的最大法拉第效率图

Fig. S13 The maximum faradaic efficiency for C₂H₄ and CH₄ as a function of N functionality content in Cu₂O/Cu@NC.

表 S1 产物浓度(x)与产物与内标物峰面积之比 $S_1 : S_2$ (y)的线性关系

Table S1 The linear fitting between the concentration of the product (x) and the ratio of product to internal standard area $S_1 : S_2$ (y).

Product	Equation
HCOOH	$y = 0.0614x + 0.0069$
CH ₃ OH	$y = 0.18x + 0.004$
CH ₃ CH ₂ OH	$y = 0.236x + 0.045$
CH ₃ COOH	$y = 0.266x - 0.02109$
CH ₃ CH ₂ CH ₂ OH	$y = 0.29x - 0.02$

表 S2 不同退火温度样品的比表面积

Table S2 Specific surface area of samples annealed at different temperature.

Sample	Specific surface area (m ² ·g ⁻¹)		
	400 °C	600 °C	800 °C
Cu ₂ O/Cu@NC	230.24	278.49	225.23
Cu ₂ O/Cu@C	109.56	41.48	16.41

表 S3 不同样品的电化学比表面积

Table S3 Electrochemical specific surface area (ECSA) of different samples.

Sample	Specific surface area (m ² ·g ⁻¹)		
	400 °C	600 °C	800 °C
Cu ₂ O/Cu@NC	1.23	0.52	0.30
Cu ₂ O/Cu@C	0.27	0.22	0.22

表 S4 不同温度下制备的 Cu₂O/Cu@NC 的原子浓度(%)

Table S4 Atomic concentration (%) of Cu₂O/Cu@NC prepared at different temperatures.

sample	Cu	N	C	O
Cu ₂ O/Cu@NC-400	2.99	9.09	70.39	17.54
Cu ₂ O/Cu@NC-600	3.58	11.42	72.24	12.76
Cu ₂ O/Cu@NC-800	6.22	5.57	71.42	16.80

表 S5 不同温度下制备的 Cu₂O/Cu@C 的原子浓度(%)

Table S5 Atomic concentration (%) of Cu₂O/Cu@C prepared at different temperatures.

sample	Cu	C	O
Cu ₂ O/Cu@C-400	2.83	80.34	16.83
Cu ₂ O/Cu@C-600	3.46	84.05	12.49
Cu ₂ O/Cu@C-800	6.09	83.95	9.96

表 S6 不同温度下制备的 Cu₂O/Cu@NC 中 Cu(0)与 Cu(I)浓度(%)

Table S6 Cu(0) and Cu(I) concentration (%) in Cu₂O/Cu@NC prepared at different temperature.

sample	Cu(I)	Cu(0)
Cu ₂ O/Cu@NC-400	62.11	37.89
Cu ₂ O/Cu@NC-600	50.76	49.24
Cu ₂ O/Cu@NC-800	35.90	64.10

表 S7 不同温度下制备的 Cu₂O/Cu@NC 的 N 物种浓度(%)

Table S7 Atomic concentration (%) of N species in Cu₂O/Cu@NC prepared at different temperatures.

sample	Pyridinic-N	Cu-N	Pyrolic-N	Graphitic-N	Quaternary-N	Oxidized-N
Cu ₂ O/Cu@NC-400	1.915	2.015	2.255	1.645	0.930	0.330
Cu ₂ O/Cu@NC-600	2.666	1.856	1.523	2.548	1.570	1.257
Cu ₂ O/Cu@NC-800	1.010	0.664	0.548	0.635	1.443	1.270